

5/6

15A

7154.

Q-688/15 RS.



Admissible under registration
Rule 21 also under section 11A (23)
of the Tenancy Act, duly
stamped under the Indian Stamp
Act 1899, Schedule I A No. 2,3

464

N 101

6/8

18/5/1954

MIDNAPUR

9.11.53

Uddhava Chandra
Bijay Krishna Chandra

Handwritten notes in the left margin, including '18/5/54' and 'N 101'.

Handwritten text in the center, appearing to be a list or record of transactions, with some lines crossed out.

11/2 23

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11/2 23

Kali Kristus Klejner
Bijou & ibna Klejner

The first part of the paper is devoted to a discussion of the
 various methods of determining the rate of reaction. The
 most common method is the measurement of the change in
 concentration of one of the reactants or products as a
 function of time. This can be done by various means, such
 as titration, gravimetry, or spectrophotometry. The rate
 of reaction is then calculated from the slope of the curve
 obtained. Other methods include the measurement of the
 change in pressure or volume of a gas, or the change in
 color of a solution. The choice of method depends on the
 nature of the reaction and the substances involved.

[Faint handwritten notes at the bottom of the page, likely bleed-through from the reverse side.]

~~Signature
of the Officer
in Charge
of the Station~~

2) Shubani Kausad
- Shubani Kausad
- Shubani Kausad

21 Sintering Basin
Chinmai Bai

84 Bhakti Bhawan Sarker
Bileiganga
Mirzapur.

Kali Mishra Majumdar
Byree & Ishwar Majumdar

17/1/2012



MIDNAPORE

9.11.53



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Vol. No. 8
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For the District Registrar
MIDNAPORE

12/11/53

